



The University of Texas at Austin  
Cockrell School of Engineering

# DESIGNING LITHIUM-ION BATTERY CATHODES

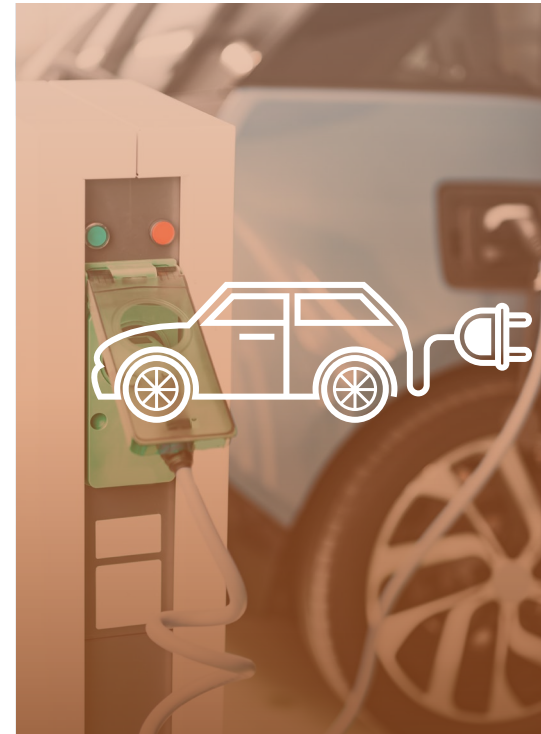
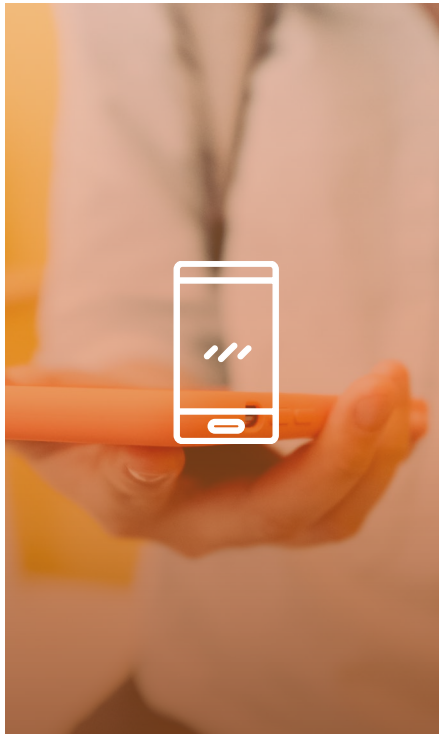
JOHN B. GOODENOUGH

Presented by  
Arumugam Manthiram  
Director, Texas Materials Institute  
The University of Texas at Austin



# LITHIUM-ION BATTERY

A DISCOVERY  
THAT CHANGED  
THE WORLD





# EARLY WORK

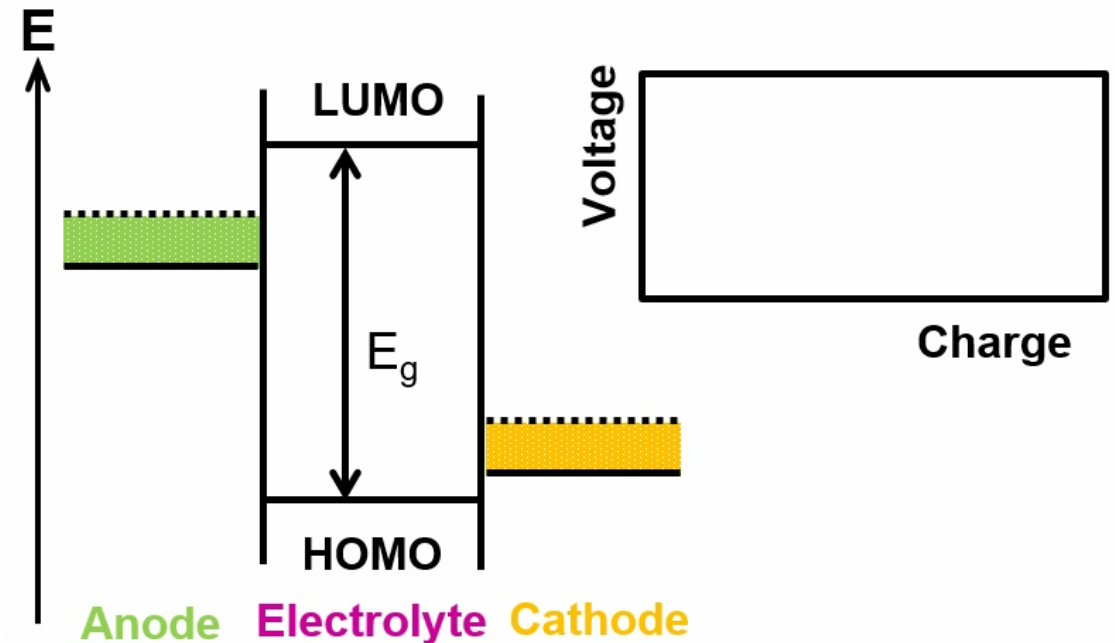
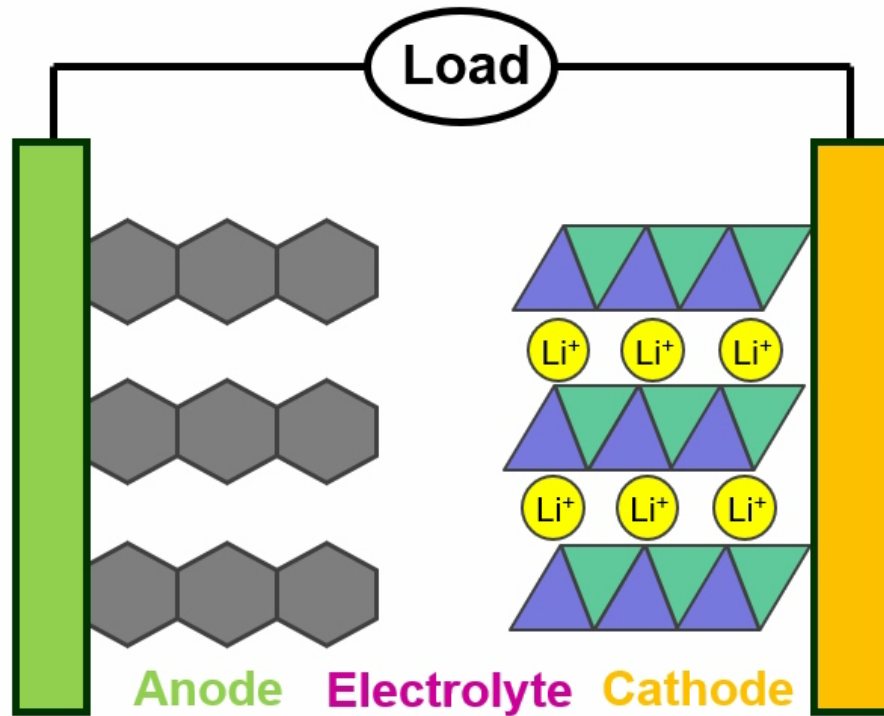
1950-1980



- Magnetic materials for first RAM memory
- Cooperative atomic orbital ordering
- Rules for sign of magnetic interactions
- Solid sodium-ion electrolyte: NASICON

# THE LITHIUM-ION BATTERY

## HOW IT WORKS



# — WHAT FACTORS DETERMINE CHOICES FOR NEW BATTERY CHEMISTRY?



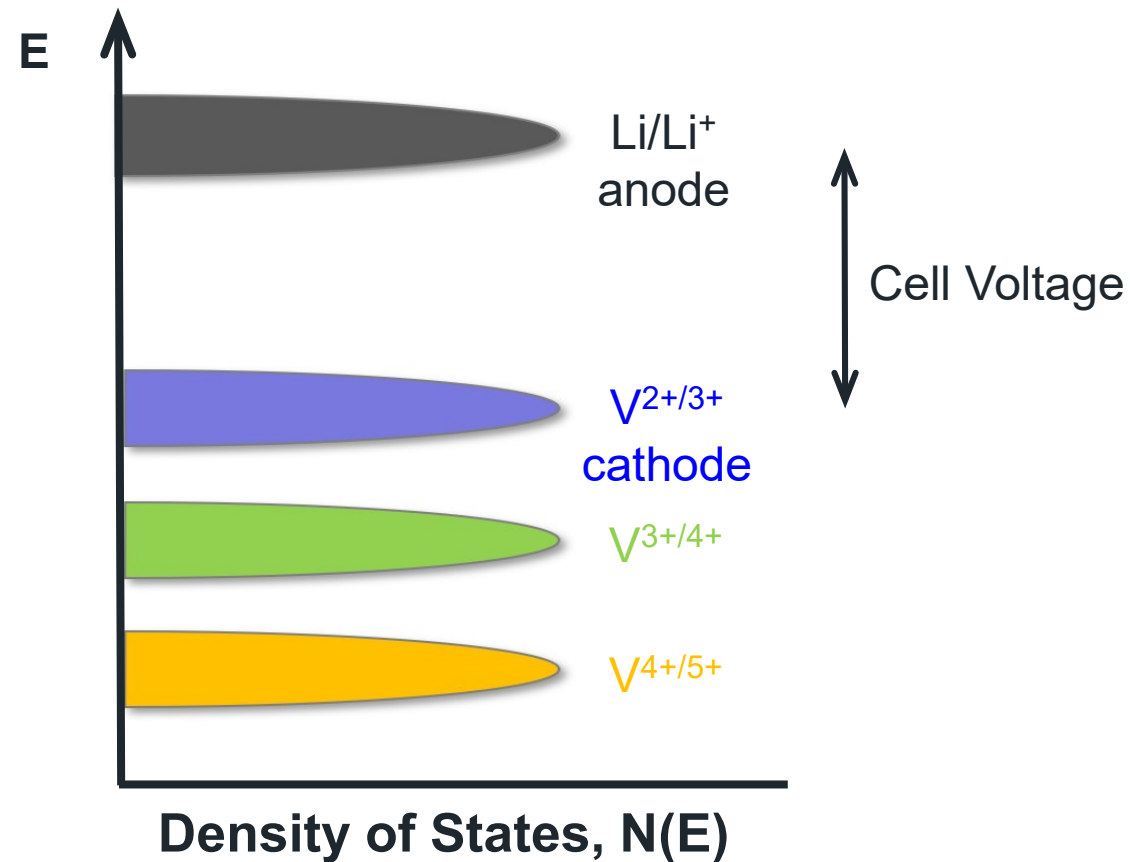
1. Cost
2. Energy
3. Power
4. Cycle Life
5. Safety
6. Environment



**MATERIALS**

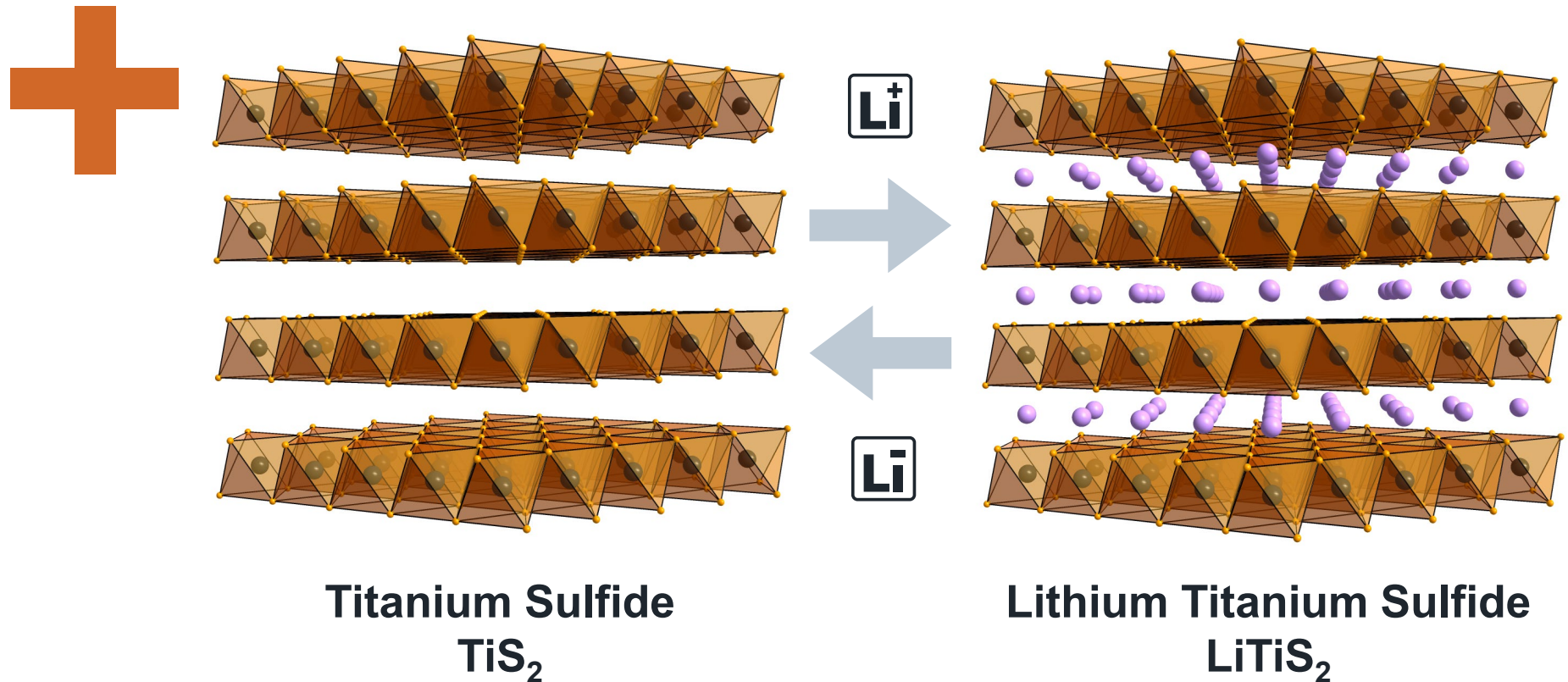
# — ENERGY DENSITY

User Time = (Cell Voltage) x (Amount of Lithium ions Stored)



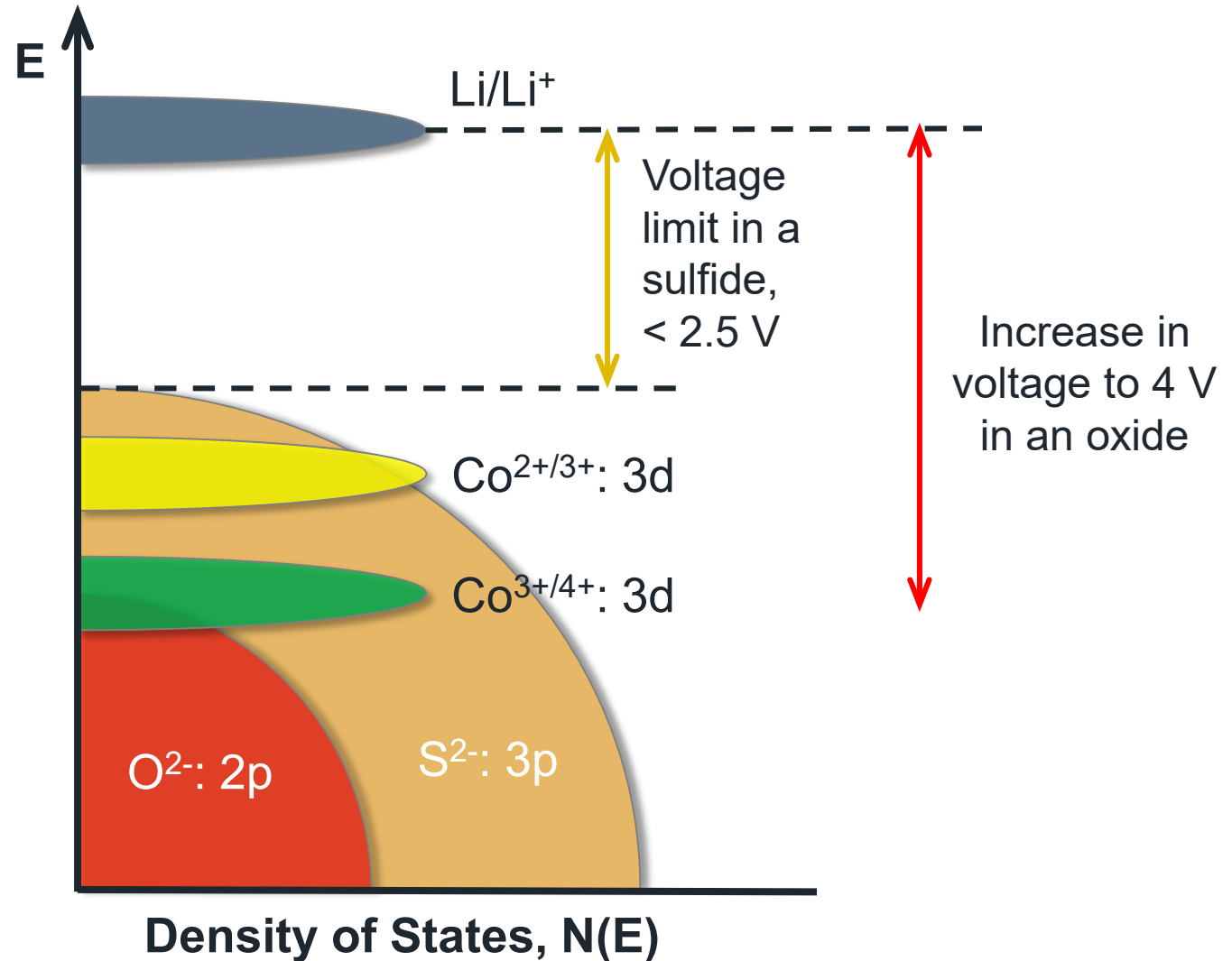
# — INSERTING LITHIUM

HOW THE  
CHEMISTRY  
WORKS



# ENERGY DENSITY

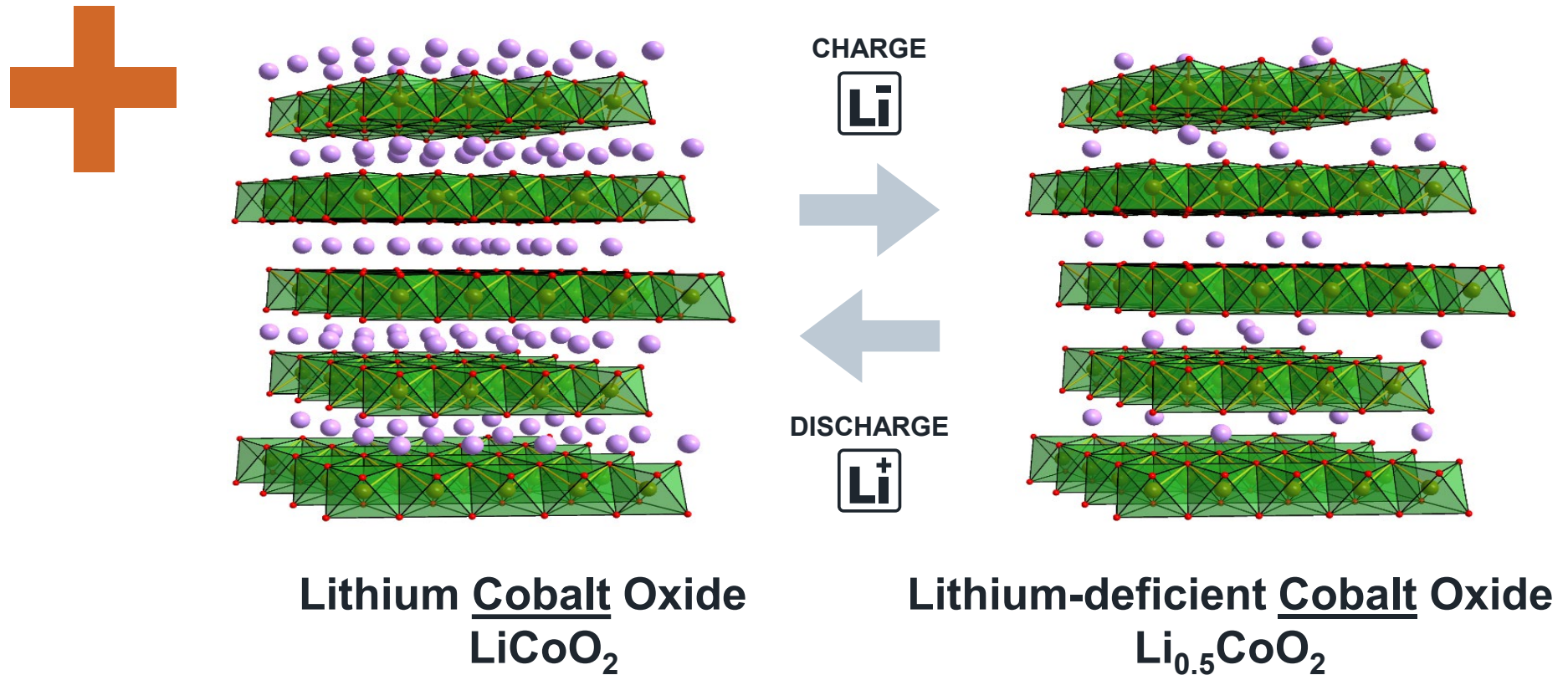
FROM SULFIDE  
TO AN OXIDE



# MATERIALS CLASS 1

## 1980: LAYERED OXIDE

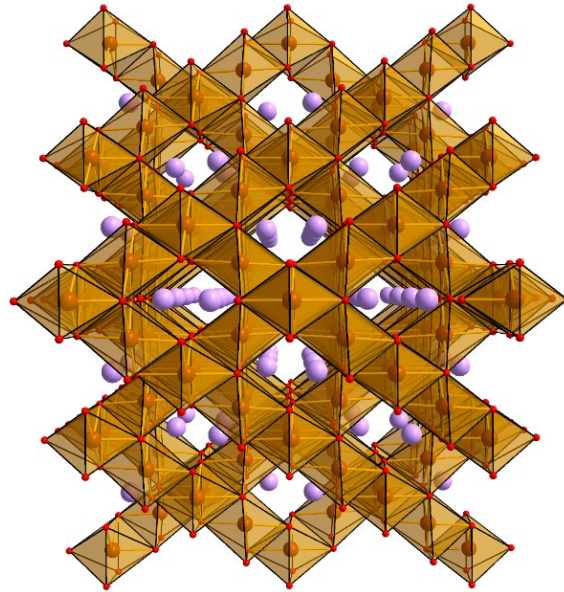
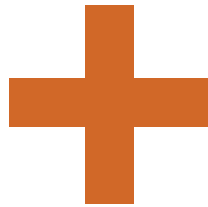
Citation: Mizushima, Jones, Wiseman, Goodenough — *Materials Research Bulletin* **15**, 783 (1980)



# MATERIALS CLASS 2

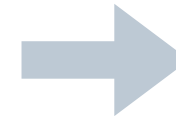
## 1983: SPINEL OXIDE

Citation: Thackeray, David, Bruce, Goodenough — *Materials Research Bulletin* **18**, 461 (1983)

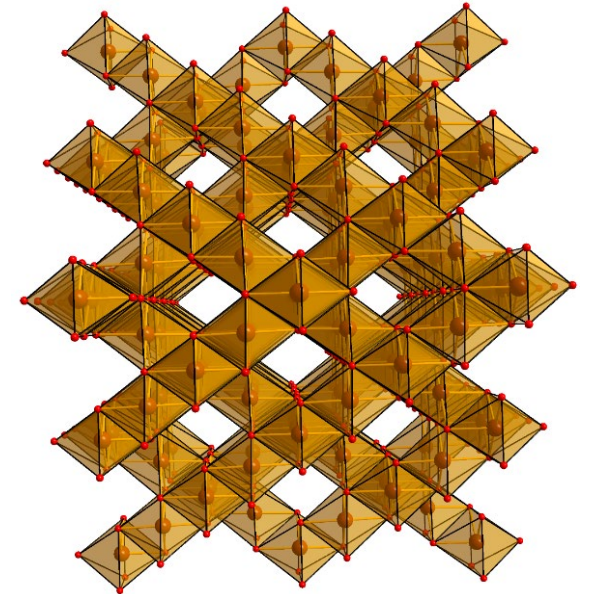


Lithium Manganese Oxide  
 $\text{LiMn}_2\text{O}_4$

CHARGE



DISCHARGE

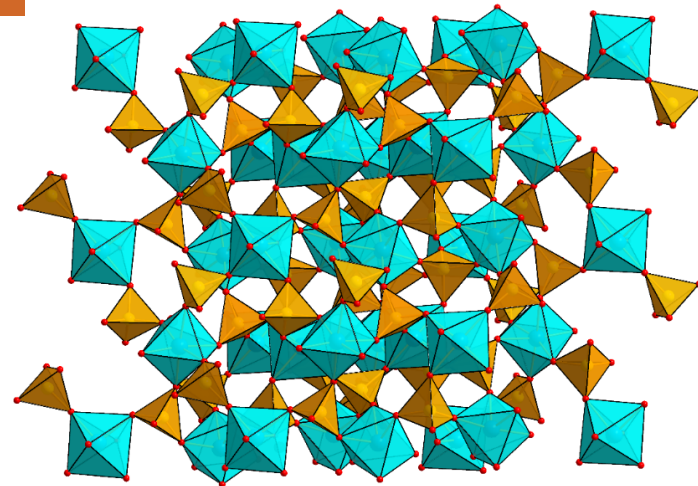


Manganese Oxide  
 $\text{Mn}_2\text{O}_4$

# MATERIALS CLASS 3

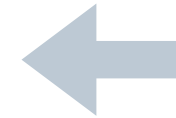
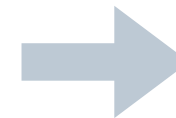
## 1987-89: POLYANION OXIDE

Citation: Manthiram, Goodenough — *Journal of Solid State Chemistry* **71**, 349 (1987)  
Manthiram, Goodenough — *Journal of Power Sources* **26**, 403 (1989)

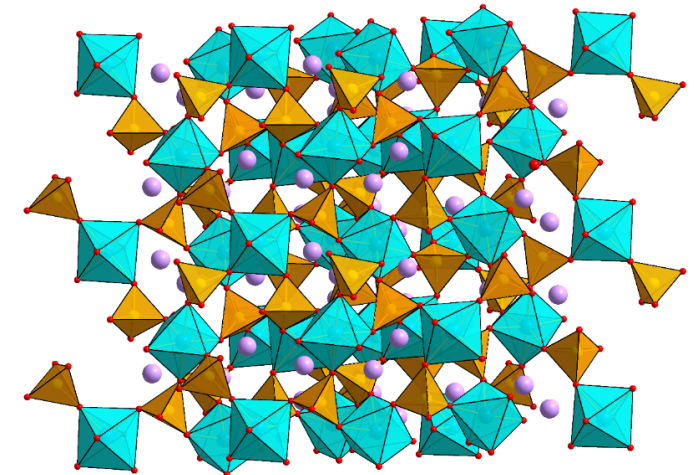


Iron Sulfate  
 $\text{Fe}_2(\text{SO}_4)_3$

DISCHARGE



CHARGE

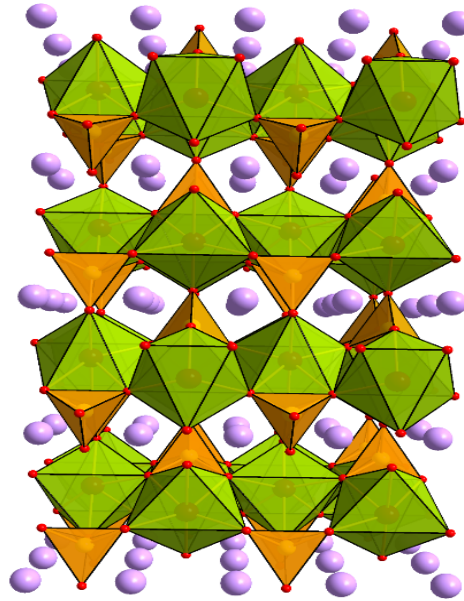
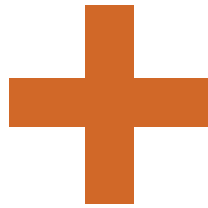


Lithium Iron Sulfate  
 $\text{Li}_2\text{Fe}_2(\text{SO}_4)_3$

# MATERIALS CLASS 3

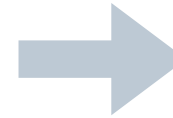
## 1997: POLYANION (OLIVINE) OXIDE

Citation: Padhi, Nanjundaswamy, Goodenough — *Journal of the Electrochemical Society* **144**, 1188 (1997)

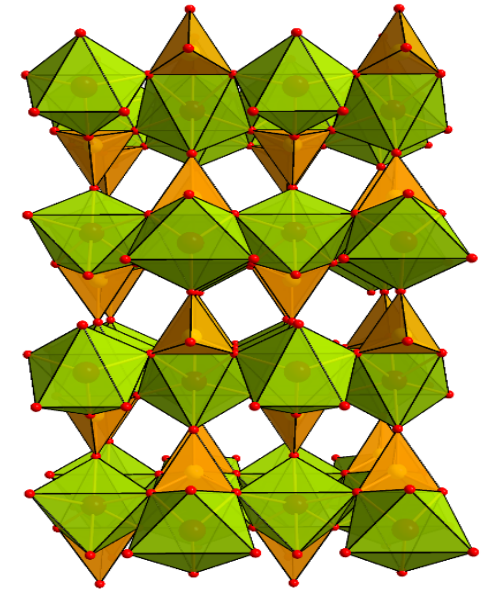


Lithium Iron Phosphate  
 $\text{LiFePO}_4$

CHARGE



DISCHARGE



Iron Phosphate  
 $\text{FePO}_4$



# KEY FINDINGS

## AND HISTORICAL SIGNIFICANCE

- A fundamental study of the properties of transition-metal oxides led to the identification of oxide cathodes
- Pushed boundaries at the intersection of solid-state chemistry and physics
- The three classes of materials discovered still remain the only viable cathodes — and the basis for future development
  - Layered oxide
  - Spinel oxide
  - Polyanion oxide

# **MOVING FORWARD**



- Liberating society from fossil fuels
- Harvesting electric power from solar and wind energy
- Electricity storage as chemical energy is the key
- Affordable, safe battery technologies



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